

## Review for Exam 2

Below you will find a list of topics that you will be responsible for knowing for Exam 2 as well as a list of topics that will not be covered. Remember that you are allowed a formula sheet for the test!

Although I've tried to cover everything, anything not explicitly mentioned is your responsibility

### Chapter 5

- De Broglie hypothesis (wave particle duality), De Broglie wavelength
- Measurement of particle wavelengths,
  - relativistic vs. non-relativistic ( $\lambda/\lambda_c$ )
- Wave packets
  - waves in general (period, frequency, wave number, phase and group velocity)
  - uncertainty:  $\Delta x \Delta k \sim 1$ ,  $\Delta \omega \Delta t \sim 1$
  - particle wave packets
- Probabilistic interpretation of wave packets
  - $P(x) dx = |\Psi(x)|^2 dx$        $\int |\Psi(x)|^2 dx = 1$
- The uncertainty principle
  - $\Delta x \Delta p \geq \hbar/2$ ,  $\Delta E \Delta t \geq \hbar/2$

Not covered: Davisson-Germer experiment, consequences of the uncertainty

### Chapter 6

- Schrodinger equation in one dimension
  - time dependent solutions
  - time independent solutions
- Infinite square well
  - solutions
  - energies
    - o ground state energy
  - probabilities
  - sketching wave functions
- Expectation values and operators
  - computation of expectation values
  - representation of operators
- Simple Harmonic Oscillator
  - potential
  - classical turning points
  - solutions (will give specific solutions on test if needed)
  - energies
    - o ground state energy
  - selection rules:  $\Delta n = \pm 1$
- reflection and transmission of waves
  - step potential

- solutions, wavenumbers
  - sketches of wave functions
- reflection and transmission coefficients

Not covered: finite square well, barrier potentials

## Chapter 7

- Schrodinger equation in three dimensions
  - solutions
  - energies
    - degeneracy
- Schrodinger equation in spherical coordinates
  - central potentials
  - separation of variables
  - solutions to spherical equation of the 3DSE
    - spherical harmonics (note: I will give specific solutions on the test)
  - quantization of angular momentum
    - angular momentum operator
    - quantum numbers l,m
    - vector representation of angular momentum
  - solution to the radial equation
    - radial functions (note: I will give specific solutions on the test)
    - energies
    - principle quantum number n
    - degeneracy
  - selection rules
- Hydrogen atom wave functions
  - normalization
  - probabilities  $P(r)dr = |\Psi|^2 4\pi r^2 dr$
  - ground and excited states, continuum states
- electron spin
  - spin quantum number
- total angular momentum
  - addition of angular momentum:  $\mathbf{J} = \mathbf{L} + \mathbf{S}$
- Pauli exclusion principle

Not covered: magnetic moments, Stern-Gerlach experiment, spin-orbit coupling, Schrodinger equation for two or more particles, ground states of atoms, excited states of atoms