

Instructor(s): Z. Qiu

PHYSICS DEPARTMENT
Exam 3

November 20, 2009

PHY 2004

Name (print, last first): _____

Signature: _____

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Specific heat of water: $c = 1 \text{ cal/g} \cdot ^\circ\text{C}$	$g = 9.80 \text{ m/s}^2$
Density of water: $\rho_w = 1 \text{ g/cm}^3$	Density of air 1.20 kg/m^3
Universal gas constant: $R = 8.314 \text{ J/mole K}$	$1 \text{ atm} = 10^5 \text{ Pa}$
Heat of fusion for water: $H_f = 80 \text{ cal/g}$	
Heat of vaporization for water: $H_v = 539 \text{ cal/g}$	

1. In order to determine the thermal conductivity of a material, one measures the heat flow through it. It is found that 1000 cal flow in 2 minutes through a 3-cm^2 area that is 0.2 cm thick. The temperature difference is maintained at 90°F . What is the k for this material (in $\text{cal/cm} \cdot \text{s} \cdot ^\circ\text{C}$)?
 - (1) 1.11×10^{-2}
 - (2) 0.82×10^{-2}
 - (3) 2.72×10^{-3}
 - (4) 3.15×10^{-2}
 - (5) 1.52×10^{-2}
2. A 500-MW electric power plant has an efficiency of 32%. It loses its waste heat in large cooling towers at 20°C . If the plant operates at the maximum possible efficiency, what are the temperature of the hot reservoir (in $^\circ\text{C}$) and the waste heat (in MJ) discharged per second respectively?
 - (1) 158, 1063
 - (2) 380, 1125
 - (3) 158, 325
 - (4) 380, 927
 - (5) 380, 355
3. The density of helium in a balloon is about 0.175 kg/m^3 . How large in volume (in m^3) must a balloon be if it is to lift a total load of 1200 kg? The 1200 kg includes the mass of the balloon but not the mass of helium in it.
 - (1) 1171
 - (2) 1025
 - (3) 976
 - (4) 1264
 - (5) 859
4. A gas is confined to a cylinder by a piston. Its original conditions are $P = 3 \text{ atm}$, $T = 20^\circ\text{C}$ and $V = V_0$. After compression, $P = 30 \text{ atm}$ and $V = V_0/6$. What is the final temperature of the gas (in $^\circ\text{C}$)?
 - (1) 215.3
 - (2) 162.5
 - (3) 291.7
 - (4) 83.9
 - (5) 127.4
5. What will be the final water temperature when 3 kg of ice at 0°C and 0.6 kg of steam at 100°C are mixed together (in $^\circ\text{C}$)?
 - (1) 40
 - (2) 30
 - (3) 20
 - (4) 50
 - (5) Not enough information given to answer.
6. A long straight pipe with internal diameter 2.50 cm is standing vertically with its lower end plugged. The pipe is filled with water up to a level so that there is 2.4 kg of water in the pipe. What is the pressure due to water on the plug at the pipe's lower end?
 - (1) 48 kPa
 - (2) 520 Pa
 - (3) 5.0 Pa
 - (4) 9.6 kPa
 - (5) 2.8 Pa

7. A 1.5 m^3 -volume tank holds oxygen gas ($M = 32 \text{ g/mole}$) at the pressure of 500,000 Pa. The temperature is 27°C . Find the mass of oxygen in the tank (in kg).
- (1) 9.62 (2) 7.45 (3) 91.2 (4) 13.24 (5) 45.7
8. A ice chest is maintained at 0°C by ice placed in it. The chest has a total surface area of 600 cm^2 and a thickness of 1 cm. The wall of the ice chest is of Styrofoam with thermal conductivity $k = 2.3 \times 10^{-4} \text{ cal}/(\text{cm s}^\circ\text{C})$. How much ice will melt in the chest in 4 hours when the outside temperature is 20°C (in g)?
- (1) 500 (2) 250 (3) 375 (4) 600 (5) none of these
9. A wood block floats on water with 35% of its volume above the surface (of water), what is the average density of the block (in kg/m^3)?
- (1) 650 (2) 720 (3) 1300 (4) 362 (5) 830
10. The filament temperature of a light bulb is 900°C when the bulb delivers 40 W of power. If its emissivity remains constant, what power (in W) is delivered when the filament temperature is 1300°C ?
- (1) 129 (2) 155 (3) 102 (4) 92 (5) 65

Instructor(s): *N. Sullivan*PHYSICS DEPARTMENT
Exam 3

November 19, 2010

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 $g = 9.80 \text{ m/s}^2$

$R = 8314 \text{ J/kmole/K}$

1. (4 points) In the compression chamber of a diesel engine the volume of a gas mixture initially at 27°C is compressed by a factor of 20 ($V_{\text{Final}} = 1/20 V_{\text{Initial}}$). If the pressure increases from 1 atmosphere to 50 atmospheres, what is the final temperature?
 - (1) 477°C
 - (2) 954°C
 - (3) 273°C
 - (4) 0°C
 - (5) 1430°C
2. (5 points) A misshapen lump of metal is weighed in air and then weighed while immersed in oil of density 800 kg/m^3 . If the weight in air is 2.2 kg and the weight in the oil is 1.2 kg, calculate the density of the metal.
 - (1) 1760 kg/m^3
 - (2) 2650 kg/m^3
 - (3) 92 kg/m^3
 - (4) 36.8 kg/m^3
 - (5) 13.5 kg/m^3
3. (4 points) A 5 meter length of steel with a cross-sectional area of 20 cm^2 is compressed with a force of 20,000 N. If the Young's modulus of steel is $200 \times 10^9 \text{ N/m}^2$, what is the change in length of the steel beam?
 - (1) 0.25 mm
 - (2) 2.5 cm
 - (3) 5.5 cm
 - (4) 0.11 cm
 - (5) 12.5 mm
4. (5 points) A cylinder of radius b has a moment of inertia $I = (1/2)mb^2$. The cylinder is rolled along a flat horizontal surface at speed v so that when it hits a ramp, it will travel to a height of 50 cm and then stop. What is the initial speed v of the cylinder?
 - (1) 2.6 m/s
 - (2) 5.25 m/s
 - (3) 12.2 m/s
 - (4) 1.22 m/s
 - (5) 52.5 m/s
5. (4 points) An object has a moment of inertia of $2.56 \text{ kg}\cdot\text{m}^2$. What is the value of the torque needed to accelerate the rotation of the object from rest to a rotation of 5 revolutions per second in 30 seconds?
 - (1) $2.68 \text{ N}\cdot\text{m}$
 - (2) $1.34 \text{ N}\cdot\text{m}$
 - (3) $0.27 \text{ N}\cdot\text{m}$
 - (4) $4.02 \text{ N}\cdot\text{m}$
 - (5) $0 \text{ N}\cdot\text{m}$
6. (4 points) A 5m^3 tank of compressed helium gas has an absolute pressure of 4 kPa at a temperature of 27°C . What is the mass of the helium in the tank? 1 kmole of helium weighs 4 kg.
 - (1) 33 g
 - (2) 13.5 g
 - (3) 1.61 kg
 - (4) 0.35 kg
 - (5) 6.42 kg
7. (4 points) A 100 meter length of steel changes temperature by 30°C during the course of a day. If the coefficient of thermal expansion of steel is 12 parts per million per $^\circ\text{C}$, what is the change in length of the steel?
 - (1) 3.6 cm
 - (2) 7.2 cm
 - (3) 14.4 mm
 - (4) 1.8 m
 - (5) 0.18 cm

Instructor(s): *N. Sullivan*PHYSICS DEPARTMENT
Midterm Exam 3

November 18, 2011

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$$g = 9.80 \text{ m/s}^2$$

$$R = 8314 \text{ J/kmole K}$$

1. (4 points) A volume of gas initially at 37°C is compressed by a factor of 10 ($V_{\text{Final}} = 1/10V_{\text{Initial}}$). If the pressure increases from 112 kPa to 1904 kPa, what is the final temperature?

- (1) 254°C (2) 855°C (3) 473°C (4) 0°C (5) 1250°C

2. (5 points) A small piece of rock is weighed in air and then weighed while immersed in oil of density 800 kg/m^3 . If the weight in air is 21.5 N and the weight in the oil is 11.8 N, calculate the density of the rock.

- (1) 1750 kg/m^3 (2) 202.7 kg/m^3 (3) 1.38 kg/m^3 (4) 13.68 kg/m^3 (5) 9710 kg/m^3

3. (4 points) A 15 meter length of steel with a cross-sectional area of 20 cm^2 is compressed with a force of 10,000 N. If the Young's modulus of steel is $200 \times 10^9 \text{ N/m}^2$, what is the change in length of the steel beam?

- (1) 0.38 mm (2) 2.3 cm (3) 4.47 mm (4) 0.19 cm (5) 12.5 mm

4. (4 points) A 10 m^3 tank of compressed natural gas has an absolute pressure of 400 kPa at a temperature of 27°C . What is the mass of the gas in the tank? 1 kmole of natural gas weighs 16 kg. $R = 8314 \text{ J/kmole K}$.

- (1) 25.6 kg (2) 288 g (3) 196 g (4) 16.1 kg (5) 642 kg

5. (4 points) A 200 meter length of steel rail changes temperature by 30°C during the course of a day. If the coefficient of thermal expansion of steel is 12 parts per million per $^\circ\text{C}$, what is the change in length of the steel?

- (1) 72 mm (2) 1.44 cm (3) 14.4 mm (4) 1.8 m (5) 3.66 mm

6. (3 points) How much water at 0°C is required to cool a 200 kg human by 1°C . The heat capacity of the human body is 3500 J/kg/K and the heat capacity of water is 4184 J/kg/K .

- (1) 167 kg (2) 239 kg (3) 83.5 kg (4) 23.9 kg (5) 200 kg

7. (3 points) An engine operating in an ideal Carnot cycle involves an isothermal compression of 2 m^3 of helium gas at 200°C and an isothermal compression of the gas at 50°C . What is the efficiency of this engine?

- (1) 31.7% (2) 75% (3) 100% (4) 63.4% (5) 87%

8. (4 points) A steel cylinder contains helium gas at a gauge pressure of 100 kPa and at a temperature of 27°C . The gas is heated and moves a piston in the cylinder 8 cm with the pressure held constant. If the area of the piston is 150 cm^2 and the mass is 7 kg, calculate the work done by the gas.

- (1) 120 J (2) 1200 J (3) 256 J (4) 2560 J (5) 0 J

Instructor(s): Z. Qiu

PHYSICS DEPARTMENT

PHY 2004

Exam 3

April 8, 2009

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Young modulus of steel:	200 GPa
Density of water:	1 g/cm ³
Specific heat of water:	$c = 1\text{ cal/g}\cdot^{\circ}\text{C}$
Specific heat of ice:	$c = 0.5\text{ cal/g}\cdot^{\circ}\text{C}$
Volume expansion coefficient of benzene:	$\beta = 1.24 \times 10^{-3} /^{\circ}\text{C}$
Universal gas constant:	$R = 8.314 \text{ J/mole K}$
1 atm = 100 kPa	1 cal = 4.18 J

1. A steel girder has a cross-sectional area of 150 cm^2 . How large a tensile force would be required to lengthen the girder by 0.01 percent (in N)?
 (1) 3.0×10^5 (2) 2.5×10^5 (3) 3.5×10^5 (4) 4.0×10^5 (5) 4.5×10^5
2. A certain piece of metal weighs 1.68 N in air and 0.82 N when submerged in water. What is the mass density of the metal (in g/cm³)?
 (1) 1.95 (2) 2.37 (3) 2.54 (4) 2.93 (5) 2.18
3. A gas is confined to a cylinder by a piston. Its original conditions are $P = 3 \text{ atm}$, $T = 20^{\circ}\text{C}$ and $V = V_0$. After compression, $P = 30 \text{ atm}$ and $V = V_0/6$. What is the final temperature of the gas (in $^{\circ}\text{C}$)?
 (1) 215.3 (2) 162.5 (3) 291.7 (4) 83.9 (5) 127.4
4. A certain amount of benzene occupies 500 cm^3 at 10°C . What will its volume be at 35°C (in cm³)?
 (1) 516 (2) 527 (3) 504 (4) 510 (5) 518
5. A 10.0 cm internal diameter pipe is standing vertically with its lower end plugged. The pipe is filled with water up to a level so that there is 19.2 kg of water in the pipe. What is the pressure of the water on the plug at the pipe's lower end?
 (1) 24 kPa (2) 240 Pa (3) 2.5 Pa (4) 4.8 Pa (5) 1.4 Pa
6. An ideal gas is confined to a vertical cylinder by a 2 kg piston. When 3 cal of heat is added to the gas, the piston lifts 30 cm. The change in internal energy of the gas is (in J)
 (1) 6.66 (2) 5.24 (3) 4.19 (4) 8.21 (5) 2.94

7. How much heat (in kcal) must be added to 3 kg of ice at -20°C in order to change it to water at 50°C ?
- (1) 420 (2) 350 (3) 620 (4) 200 (5) 160
8. An ideal Carnot engine operates between two reservoirs which are at the temperatures 100°C and 200°C . For each 100 J heat energy taken in, how much heat (in J) is exhausted?
- (1) 78.9 (2) 50.0 (3) 21.1 (4) 73.2 (5) 26.8
9. Gas in a 100 cm^3 container has an absolute pressure $3.3 \times 10^5\text{ Pa}$ at 20°C . What absolute pressure will it have at 120°C ?
- (1) 4.4×10^5 (2) 1.3×10^5 (3) 3.1×10^5 (4) 6.8×10^3 (5) 6.4×10^5
10. The filament temperature of a light bulb is 2000 K when the bulb delivers 40 W of power. If its emissivity remains constant, what power (in W) is delivered when the filament temperature is 2500 K?
- (1) 98 (2) 105 (3) 62 (4) 50 (5) 45

Instructor(s): Z. Qiu

PHYSICS DEPARTMENT
Exam 3

April 7, 2010

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Young modulus of steel	200 GPa
Density of water	1 g/cm ³
Volume expansion coefficient of benzene	$\beta = 1.24 \times 10^{-3}/^{\circ}\text{C}$
Specific heat of water	$c = 1 \text{ cal}/g \cdot ^{\circ}\text{C}$
Heat of fusion for water	$H_f = 80 \text{ cal}/g$
Heat of vaporization for water	$H_v = 539 \text{ cal}/g$
1 atm = 100 kPa	

1. A steel girder has a cross-sectional area of 50 cm². How large a tensile force would be required to lengthen the girder by 0.03 percent (in N)?
 (1) 3.0×10^5 (2) 2.5×10^5 (3) 3.5×10^5 (4) 4.0×10^5 (5) 4.5×10^5
2. A certain piece of metal weighs 1.68 N in air and 0.77 N when submerged in water. What is the mass density of the metal (in g/cm³)?
 (1) 1.85 (2) 2.37 (3) 2.54 (4) 2.93 (5) 2.18
3. A gas is confined to a cylinder by a piston. Its original conditions are $P = 3 \text{ atm}$, $T = 20^{\circ}\text{C}$ and $V = V_0$. After compression, $P = 30 \text{ atm}$ and $V = V_0/6$. What is the final temperature of the gas (in $^{\circ}\text{C}$)?
 (1) 215.3 (2) 162.5 (3) 291.7 (4) 83.9 (5) 127.4
4. A certain amount of benzene occupies 500 cm³ at 10 $^{\circ}\text{C}$. What will its volume be at 35 $^{\circ}\text{C}$ (in cm³)?
 (1) 516 (2) 527 (3) 504 (4) 510 (5) 518
5. A 2.50 cm internal diameter pipe is standing vertically with its lower end plugged. The pipe is filled with water up to a level so that there is 1200 g of water in the pipe. What is the pressure of the water on the plug at the pipe's lower end?
 (1) 24 kPa (2) 240 Pa (3) 2.5 Pa (4) 4.8 Pa (5) 1.4 Pa

6. In order to determine the thermal conductivity of a material, one measures the heat flow through it. It is found that 1000 cal flow in 2 minutes through a 3-cm^2 area that is 0.2 cm thick. The temperature difference is maintained at 90°F . What is the k for this material (in $\text{cal/cm}\cdot\text{s}\cdot^\circ\text{C}$)?
- (1) 1.11×10^{-2} (2) 0.82×10^{-2} (3) 2.72×10^{-3} (4) 3.15×10^{-2} (5) 1.52×10^{-2}
7. A 500-MW electric power plant has an efficiency of 32%. It loses its waste heat in large cooling towers at 20°C . If the plant operates at the maximum possible efficiency, what are the temperature of the hot reservoir (in $^\circ\text{C}$) and the waste heat (in MJ) discharged per second respectively?
- (1) 158, 1063 (2) 380, 1125 (3) 158, 325 (4) 380, 927 (5) 380, 355
8. What will be the final water temperature when 3 kg of ice at 0°C and 0.6 kg of steam at 100°C are mixed together (in $^\circ\text{C}$)?
- (1) 40 (2) 30 (3) 20 (4) 50 (5) Not enough information given to answer.
9. The filament temperature of a light bulb is 900°C when the bulb delivers 40 W of power. If its emissivity remains constant, what power (in W) is delivered when the filament temperature is 1300°C ?
- (1) 129 (2) 155 (3) 102 (4) 92 (5) 65
10. An ideal gas is confined to a vertical cylinder by a 3 kg piston. When 3 cal of heat is added to the gas, the piston lifts 20 cm. The change in internal energy of the gas is (in J)
- (1) 6.67 (2) 5.24 (3) 4.19 (4) 8.21 (5) 2.94

Instructor(s): *N. Sullivan*PHYSICS DEPARTMENT
Exam 3

PHY 2004

April 6, 2011

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$$g = 9.80 \text{ m/s}^2$$

$$R = 8314 \text{ J/kmole/K}$$

1. (4 points) A volume of a gas mixture initially at 27°C is compressed by a factor of 20 ($V_{\text{Final}} = 1/20V_{\text{Initial}}$). If the pressure increases from 1 atmosphere to 40 atmospheres, what is the final temperature?
 - (1) 327°C
 - (2) 954°C
 - (3) 273°C
 - (4) 0°C
 - (5) 1430°C
2. (6 points) A misshapen lump of metal is weighed in air and then weighed while immersed in oil of density 800 kg/m^3 . If the weight in air is 22 g and the weight in the oil is 12 g, calculate the density of the metal.
 - (1) 1760 kg/m^3
 - (2) 2650 kg/m^3
 - (3) 92 kg/m^3
 - (4) 36.8 kg/m^3
 - (5) 13.5 kg/m^3
3. (3 points) A 1 meter length of solid rubber has a circular cross-sectional area of 1 cm^2 . If the rubber is stretched with a force of 10 N, what is the total increase in length if the Young's modulus of rubber is 10^6 Pa ?
 - (1) 10 cm
 - (2) 2 mm
 - (3) 0.1 mm
 - (4) 10 mm
 - (5) 0.01 mm
4. (5 points) A cylinder of radius b has a moment of inertia $I = (1/2)mb^2$ where m is the mass. The cylinder is rolled along a flat horizontal surface at speed v so that when it hits a ramp, it will travel to a height of 50 cm and then stop. What is the initial speed v of the cylinder?
 - (1) 2.6 m/s
 - (2) 5.25 m/s
 - (3) 12.2 m/s
 - (4) 1.22 m/s
 - (5) 52.5 m/s
5. (4 points) An object has a moment of inertia of $5.12 \text{ kg}\cdot\text{m}^2$. What is the value of the torque needed to accelerate the rotation of the object from rest to a rotation of 5 revolutions per second in 30 seconds?
 - (1) $5.36 \text{ N}\cdot\text{m}$
 - (2) $1.34 \text{ N}\cdot\text{m}$
 - (3) $0.27 \text{ N}\cdot\text{m}$
 - (4) $14.2 \text{ N}\cdot\text{m}$
 - (5) $0 \text{ N}\cdot\text{m}$
6. (4 points) A 5 m^3 tank of compressed natural gas has an absolute pressure of 4 kPa at a temperature of 27°C . What is the mass of the natural gas in the tank? 1 kmole of natural gas weighs 16 kg.
 - (1) 0.128 kg
 - (2) 36.5 kg
 - (3) 1.61 kg
 - (4) 1.3 g
 - (5) 6.42 kg
7. (4 points) A 10 meter length of steel changes temperature by 20°C during the course of a day. If the coefficient of thermal expansion of steel is 12 parts per million per $^\circ\text{C}$, what is the change in length of the steel?
 - (1) 2.4 mm
 - (2) 4.8 cm
 - (3) 28.8 mm
 - (4) 1.8 m
 - (5) 0.82 mm