

PHZ3113–Introduction to Theoretical Physics

Fall 2008

Problem Set 5

September 17, 2008

Due: Friday, Sept. 19, 2008

Reading: Boas chapt. 4

1. Find

$$\frac{d}{dx} \int_1^{1/x} \frac{e^{xt}}{t} dt \quad (1)$$

2. The equation of state for a van der Waals gas is

$$\left(p + \frac{a}{v^2}\right) (v - b) = C. \quad (2)$$

Calculate dv/dp by the method of total differentials.

3. Expand

(a) $f(x, y) = xy^2 + 2x^2 + 1$ around $(1, 2)$.

(b) $g(x, y) = \frac{y^2}{x^3}$ around $(1, 0)$.

4. Minimize the function $f = x^2 + y^2$ subject to the constraint $y = 1 - x^2$.

5. Problem 5 (Shankar) on next page

Problem 3.1.7. A Statistical Mechanics Interlude. This example shows how Lagrange multipliers appear in Statistical Mechanics. Consider N particles in a box. According to quantum mechanics, the energies of the particles are quantized to some set of values or energy levels, $\epsilon_1, \epsilon_2, \dots$. Let n_i be the number of particles in level i with energy ϵ_i . The multiplicity or number of distinct rearrangements of the particles consistent with any given distribution n_i , is given by

$$W(n_1, n_2, \dots) = \frac{N!}{n_1! n_2! \dots} \quad (3.1.61)$$

For example if all the particles are in the lowest energy level, we have $n_1 = N$, rest of the $n_i = 0$ and $W = 1$. (Recall that $0! = 1$.) If one particle is in the second level and the rest are still in the lowest, $W = N!/(N-1)! = N$, the multiplicity reflecting the N ways to choose the one who goes up. The question is this: which distribution of particles, subject to the constraint that the total number equal N and the total energy equal E , gives the biggest W ? Proceed to find this as follows:

- a) • Work with $S = \ln W$. Argue that

$$\ln n! \simeq n \ln n - n$$

for large n by approximating the sum involved in $\ln n!$ by an integral.

- b) • Write the constraints on the n_i 's due to total number N and energy E .
- c) • Treat all n_i as continuous variables, introduce Lagrange multipliers α and β for N and E and maximize S .
- d) • Derive the Boltzmann distribution $n_i = e^{-\alpha - \beta \epsilon_i}$.

The multipliers may then be found by setting the total number and energy coming from this distribution to N and E , respectively. But this is not our problem here.