

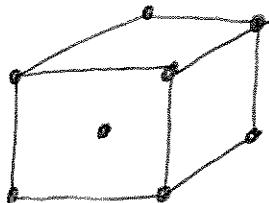
Crystal Lattices - Chpt 4

Bravais Lattice: $\{\vec{R} = n_1 \vec{a}_1 + n_2 \vec{a}_2 + n_3 \vec{a}_3\}$
 \vec{a} 's = basis

Examples:

Simple cubic: $a\hat{x}, a\hat{y}, a\hat{z}$

Body Centered Cubic: (BCC)



$$a\hat{x}, a\hat{y}, \frac{a}{2}(\hat{x} + \hat{y} + \hat{z})$$

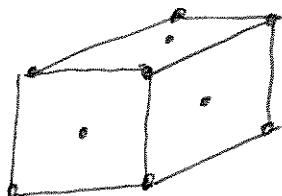
More symmetric:

$$\vec{a}_1 = \frac{a}{2}(\hat{y} + \hat{z} - \hat{x})$$

$$\vec{a}_2 = \frac{a}{2}(\hat{z} + \hat{x} - \hat{y})$$

$$\vec{a}_3 = \frac{a}{2}(\hat{x} + \hat{y} - \hat{z})$$

Face Centered Cubic: (FCC)



$$\vec{a}_1 = \frac{a}{2}(\hat{y} + \hat{z})$$

$$\vec{a}_2 = \frac{a}{2}(\hat{z} + \hat{x})$$

$$\vec{a}_3 = \frac{a}{2}(\hat{x} + \hat{y})$$

Coordination number = number of nearest neighbors

6, 8, 12 above

2.

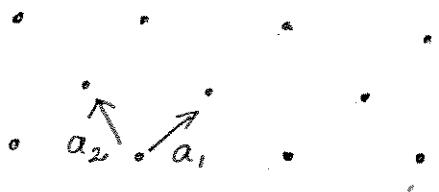
Primitive Unit Cell:

Volume when translated by \vec{R} fills space w/out overlapping. (not unique)

Unit cell:

Fills space when translated by a subset of \vec{R} .

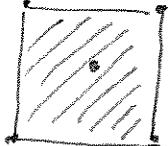
2D example:



Primitive unit cell

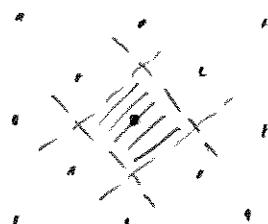


Unit cell:



Wigner-Seitz primitive cell:

Points closest to a given lattice point:

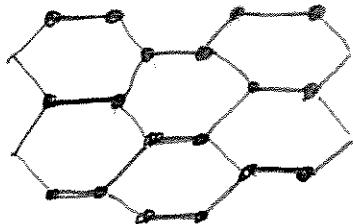


Show 3-D examples.

Lattice w/a basis: $\{\vec{R} + \sum_i \vec{b}_i\}$

↑
lattice
basis

Examples:



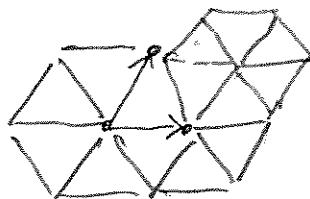
BCC: 2 point basis: $\vec{0}, \frac{a}{2}(\hat{x} + \hat{y} + \hat{z})$

FCC: 4 point basis: $\vec{0}, \frac{a}{2}(\hat{x} + \hat{y}), \frac{a}{2}(\hat{y} + \hat{z}), \frac{a}{2}(\hat{z} + \hat{x})$

Diamond: FCC w/ $\vec{b}_i = \vec{0}, \frac{a}{4}(\hat{x} + \hat{y} + \hat{z})$

Hexagonal Close-Packed (HCP)

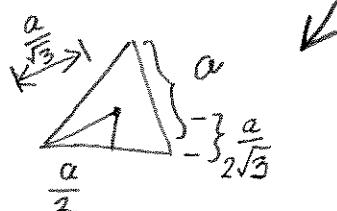
$$a_1 = a\hat{x}, a_2 = \frac{a}{2}\hat{x} + \frac{\sqrt{3}}{2}a\hat{y}, a_3 = c\hat{z} \text{ (simple hexagonal)}$$



$$\text{basis} = \vec{0}, \frac{1}{3}\vec{a}_1 + \frac{1}{3}\vec{a}_2 + \frac{1}{3}\vec{a}_3$$

$$c = \sqrt{\frac{8}{3}}a \quad \text{Why?}$$

• 1 layer
 $r = \frac{a}{2}$



\therefore Next layer is shifted by: } $\left\{ \frac{a}{\sqrt{3}} + \frac{a}{2\sqrt{3}} = \frac{3}{2} \frac{a}{\sqrt{3}} = \frac{\sqrt{3}}{2}a\sqrt{3} \right.$

$$\left. \frac{a}{\sqrt{3}} \left(\frac{\sqrt{3}}{2}\hat{x} + \frac{1}{2}\hat{y} \right) = \frac{a}{2}\hat{x} + \frac{a}{2\sqrt{3}}\hat{y} \right\}$$

$$= \frac{1}{3} (\vec{a}_1 + \vec{a}_2)$$

Third layer shifts back over first layer in HCP.



$$a^2 - \frac{a^2}{3} = h^2 \rightarrow h = \frac{\sqrt{2}}{\sqrt{3}} a$$

(tetrahedron)

Since there are two layers,

$$c = 2h = \frac{\sqrt{8}}{3} a.$$

We could have also continued to shift (ABC vs. AB). This creates a lattice w/ basis vectors

$$\overrightarrow{a_1} = \frac{a\hat{x}}{2}, \quad \overrightarrow{a_2} = \underbrace{\frac{a}{2}\hat{x} + \frac{\sqrt{3}}{2}a\hat{y}}, \quad \overrightarrow{a_3} = \underbrace{\frac{a}{2}\hat{x} + \frac{a}{2\sqrt{3}}\hat{y} + \frac{\sqrt{2}}{\sqrt{3}}a\hat{z}}.$$

Note that $|\vec{a}_1| = |\vec{a}_2| = a$ and $|\vec{a}_3|^2 = a^2 \left(\frac{1}{4} + \frac{1}{12} + \frac{2}{3}\right) = a^2$, and $\vec{a}_1 \cdot \vec{a}_2 = \frac{a^2}{2}$

$$\vec{a}_2 \cdot \vec{a}_3 = \frac{a^2}{4} + \frac{a^2}{4} = \frac{a^2}{2}.$$

$$\vec{a}_3 \cdot \vec{a}_1 = \frac{a^2}{2}$$

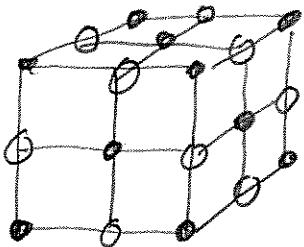
Thus, the vectors each make an angle w/ each other of $\cos^{-1}(\frac{1}{2})$. For an FCC lattice we also have

$$\frac{\frac{a}{2}(\hat{y} + \hat{z}) \cdot \frac{a}{2}(\hat{z} + \hat{x})}{\left(\frac{a}{2}\right)^2 \cdot 2} = \frac{1}{2}.$$

Sodium Chloride

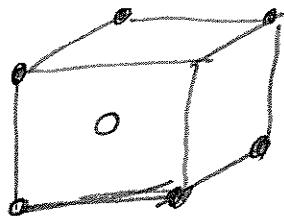
FCC w/basis

$$\vec{O}, \frac{a}{2}(\hat{x} + \hat{y} + \hat{z})$$

Cesium Chloride

SC w/basis

$$\vec{O}, \frac{a}{2}(\hat{x} + \hat{y} + \hat{z})$$

Zincblende Structure

Diamond w/alternating atoms.

Diamond is already a lattice w/basis.