

16. XRO

XRO—X-ray optics—calculates the reflectance, reflectance phase, thin-film transmittance, conductivity, dielectric function, and complex refractive index at X-ray wavelengths.

The scattering functions f_1 and f_2 for atomic constituents are those of Henke (1993)*, with new KK transforms for f_1 . The data are embedded in the `xro.exe` file; auxiliary files are not needed.

You will need to specify the formula, e. g., `YBa2Cu3O6.93`. Fractional subscripts are fine. Case matters: cobalt (Co) is not carbon monoxide (CO)! XRO does not do very complicated parsing of formulae, so a material with a formula of $(\text{La}_{0.5}\text{Pr}_{0.5})_{0.67}\text{Ca}_{0.33}\text{MnO}_3$ would be reported as `La0.335Pr0.335Ca0.33MnO3`.

You will also need to specify either V , the volume per formula unit in cubic Angstroms, or the density, in grams/cc. Here, V is the volume of the formula unit, typically V_{cell}/Z as specified by the structural paper. Z is the number of formula units in the unit cell. For example, NaCl is fcc with a lattice constant of 5.64 Å. The unit cell volume is 179.4 Å³ but $Z = 4$, so $V = 44.8$ Å³. Note that density always works. You get the same xro data using the density of 2.165 g/cm³ for NaCl or Na₄Cl₄. The volumes V would be 44.8 and 179.4, respectively.

Examples:

`XRO YBa2Cu3O6.93 /v186`

`XRO silver /fAg /r10.49 /P`

In the first, XRO calculates for YBCO, using a cell volume of 186 Å³. In the second the reflectance and the phase are computed for Ag, based on a density of 10.49 g/cm³. The output file is silver.XRP.

Switches:

<code>/F<formula></code>	— Chemical formula
<code>/Vnnn</code>	— Volume/formula unit is <i>nnn</i>
<code>/Dnnn /Rnnn</code>	— Density is <i>nnn</i>
<code>/E</code>	— Energy in eV rather than cm ⁻¹
<code>/P</code>	— Write also the phase; 3 columns data
<code>/S1</code>	— Write conductivity, $\sigma_1(\omega)$
<code>/E1</code>	— Write dielectric const, $\epsilon_1(\omega)$
<code>/I</code>	— Write Index (n and κ) rather than \mathcal{R} and (with \P) ϕ
<code>/T</code>	— Write Transmittance, \mathcal{T}
<code>/Xnnn</code>	— For \mathcal{T} : the thickness (in Å) is <i>nnn</i>
<code>/F</code>	— Write f_1 and f_2 (for the <i>first</i> atom in the formula)
<code>/q</code>	— Fewer questions

Switches may be introduced with ‘/’ or ‘-’, have optional spaces between them, and may be in upper or lower case.

* B.L. Henke, E.M. Gullikson, and J.C Davis, “X-ray Interactions: Photoabsorption, Scattering, Transmission, and Reflection at E=50–30,000 eV, Z=1–92,” *Atomic Data and Nuclear Data Tables* **54**, 181–342 (1993).

You may enter the formula on command line, as in `XRO YBa2Cu3O6.93 ...` or using the `/F` switch. The output file is `<formula>.XRO`. If you enter a non-formula on the command line, that will be the name of the output file and you will need to use the `/F` switch or answer the prompt for the formula. (Be careful here! YBCO is shorthand for the 123 superconductor, but it also is yttrium boron carbon oxide. Lower case, e. g., `ybco`, is always safe.)

XRO calculates the dielectric function using the original atomic scattering factor data developed by Henke. Some information about this is at http://henke.lbl.gov/optical_constants/.*

The dielectric function[†] is

$$\epsilon = 1 - \sum_j \frac{4\pi n_j e^2}{m\omega^2} (f_1^j - i f_2^j) \quad (74)$$

where the sum runs over atoms j at number density n_j and with complex scattering factor f^j . Note that this has the right limiting high-frequency behavior, because $f_1^j \rightarrow Z^j$ (with Z^j here the atomic number)[‡] and $f_2 \rightarrow 0$, so that $\epsilon \rightarrow 1 - \sum_j 4\pi n_j Z^j e^2 / m\omega^2$

The refractive index is $N = \sqrt{\epsilon}$ and the reflectance is calculated from the usual equation, Eq. 57. The transmittance is calculated from Eqs. 59–64.

* XRO uses the first version of the scattering data because it is sampled at identical photon energies for each atom; the newer version is not.

† Henke—and the website above—write an equation for the refractive index $N = 1 - (nr_0\lambda^2/2\pi)(f_1 + i f_2)$ with $r_0 = e^2/mc^2$ the classical radius of the electron and $\lambda = 2\pi c/\omega$ the wavelength. This is clearly an expansion of $N = \sqrt{\epsilon}$.

‡ There is a small relativistic correction.